


Key technologies



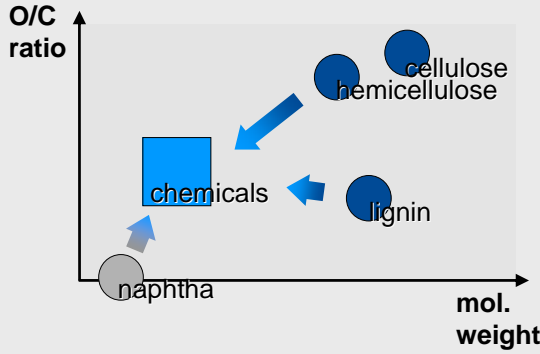
BASF
The Chemical Company

Biomass:

- a mixture of highly functionalised chemicals
- low transport density
- low energy density

Issues:

- solid handling
- fractionation
- dilute solutions
- defunctionalisation




⇒ **Key technologies:**

- refinement
- catalysis (chemical & biotechnological)

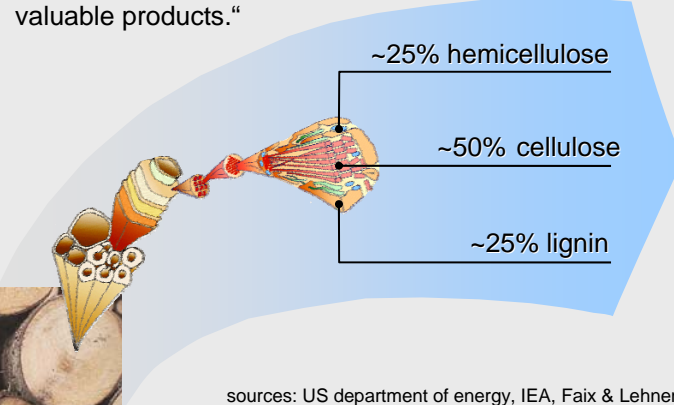
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Biorefinery technology



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“A biorefinery is an overall concept of a processing plant, where lignocellulosic **biomass** feedstocks are converted and extracted into a **spectrum** of valuable products.“



~25% hemicellulose

~50% cellulose

~25% lignin

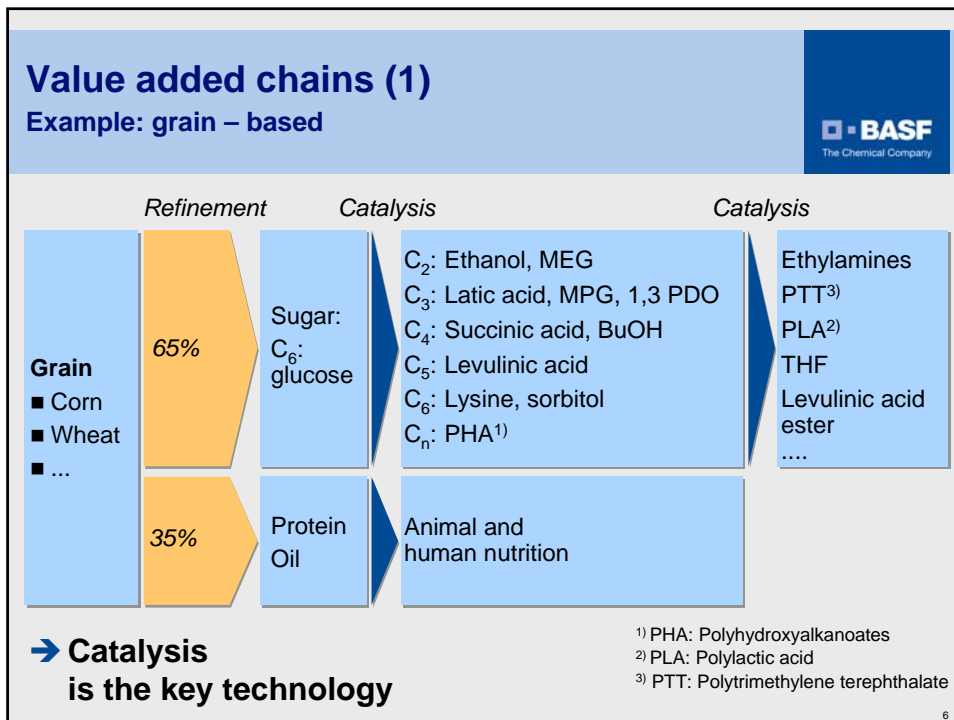
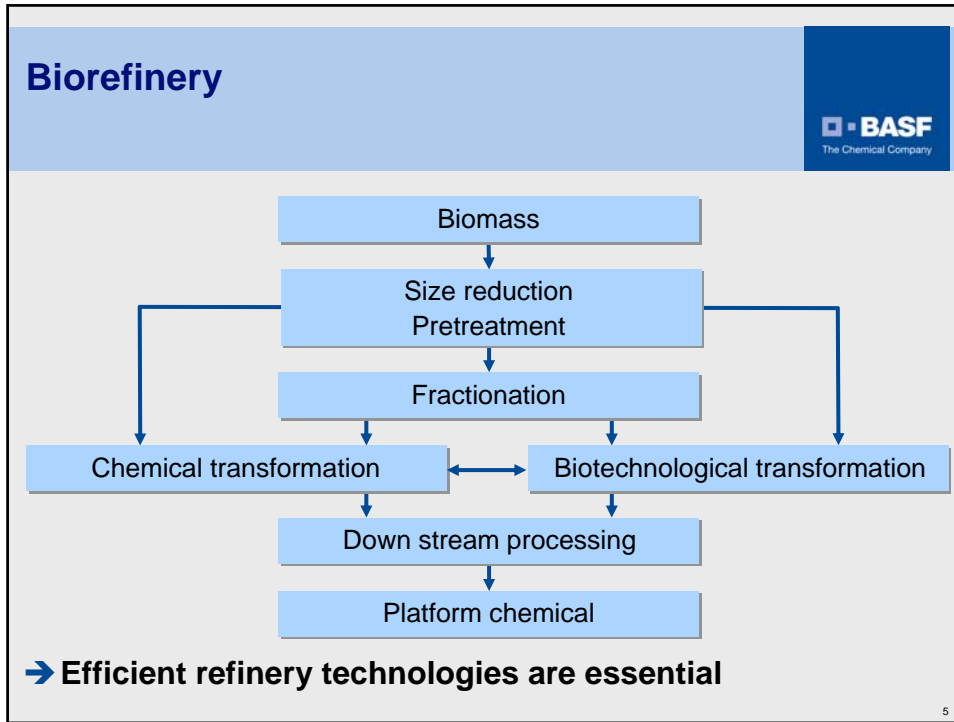
biofuels

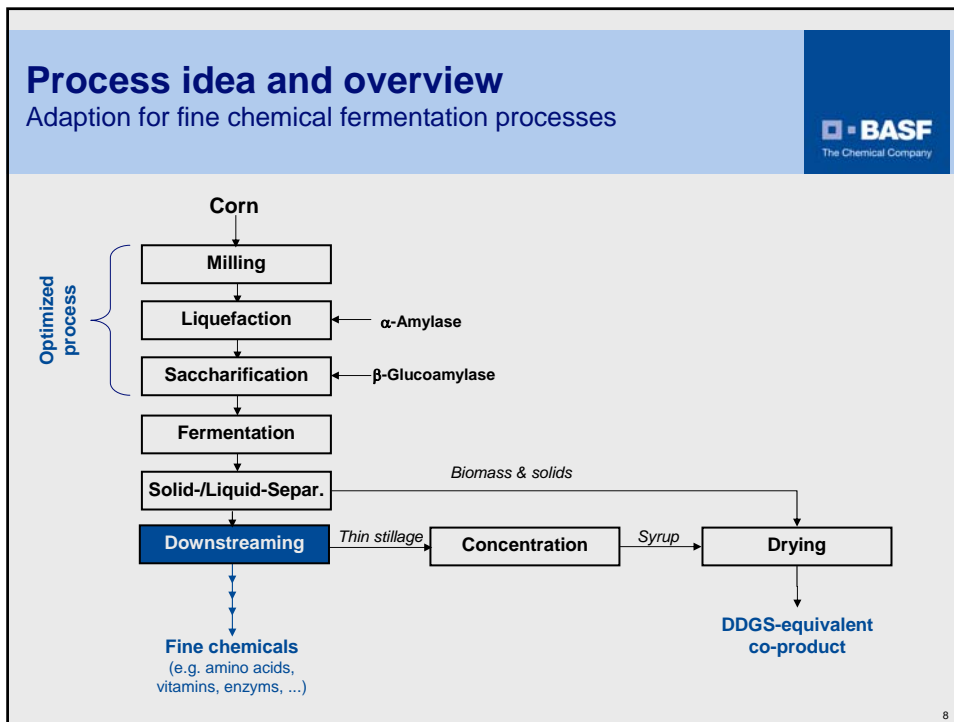
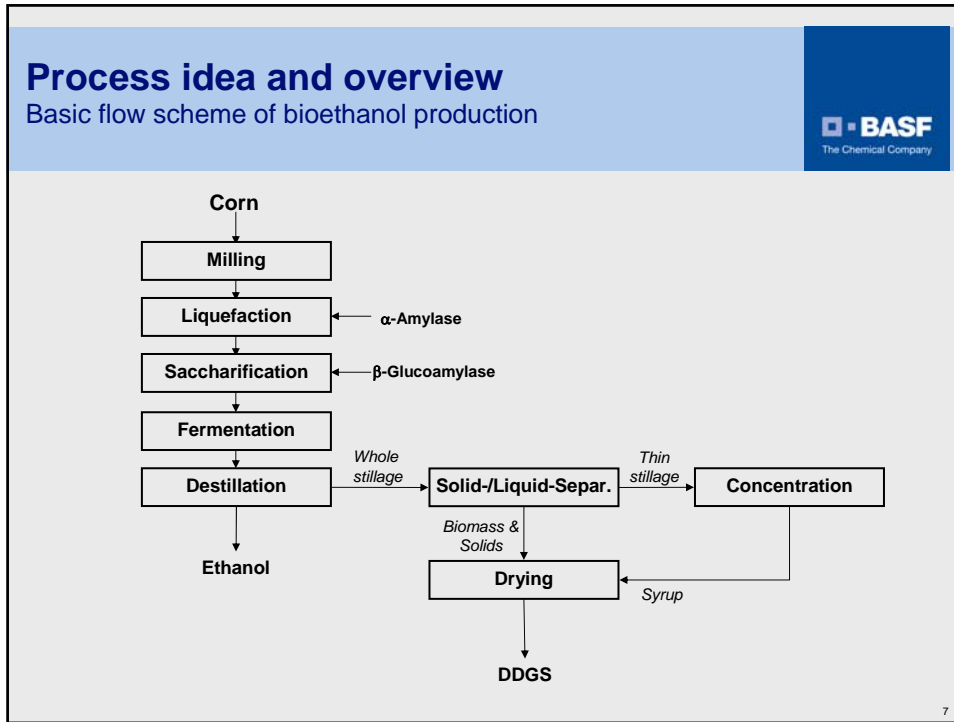
energy

chemicals

sources: US department of energy, IEA, Faix & Lehnen

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Process idea and overview

Pilot-scale development





Overall view pilot plant

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Fermentation sugar sources

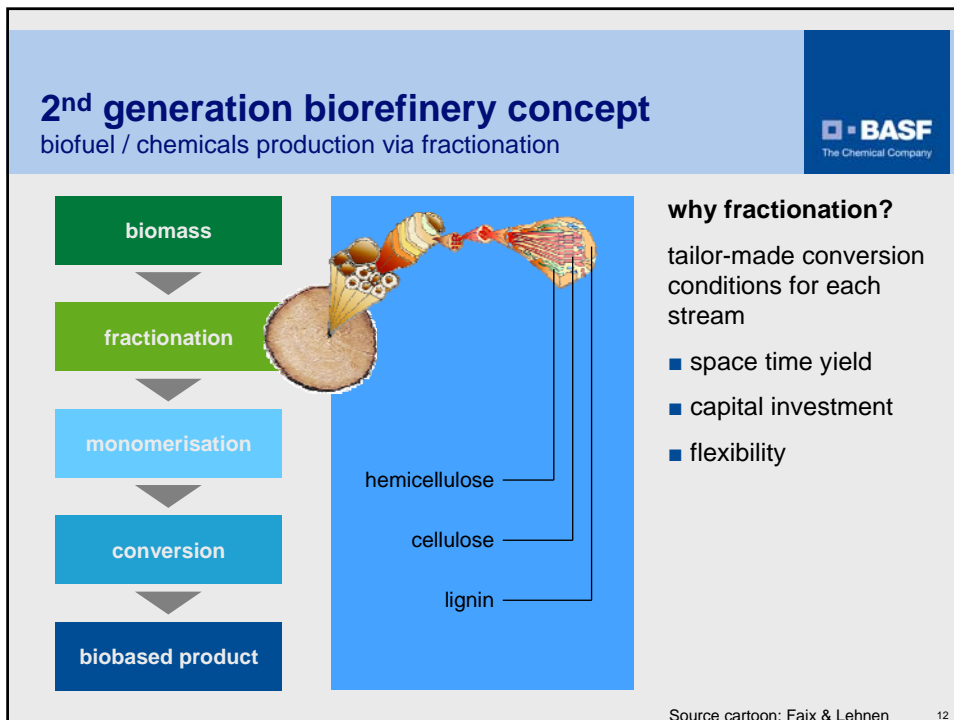
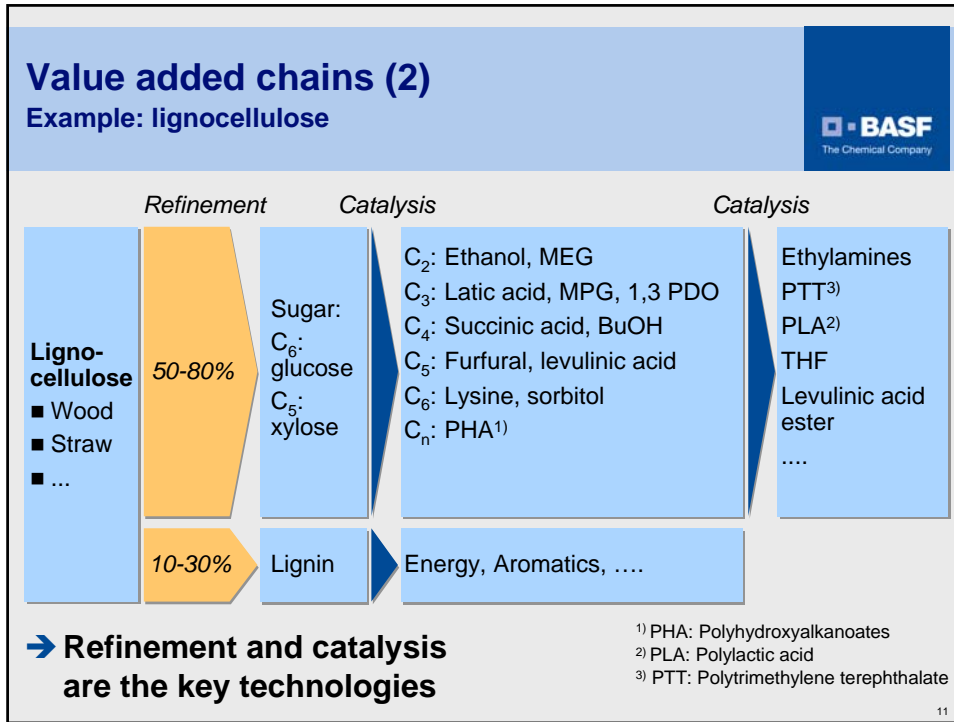
Comparison of existing technologies & BASF process



Process	Sugar mill	Wet-milling	Dry-milling (BioEtOH)	BASF process
Raw material	Sugar cane	Corn	Corn	Corn
Fermentation sugar purity	> 98 %	> 99% (food-grade)	~ 70%	> 90%
Autonomy of sugar production	Low	Low	High	High
Raw material costs	World market	World market	World market	World market
Investment costs	Low	High ¹	Low	Medium
Production costs	Low	Low ¹	Low	Low


¹ World-scale plants (>1.5 Mio tons/a crushing capacity)


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Example: biomass pretreatment

ionic liquids






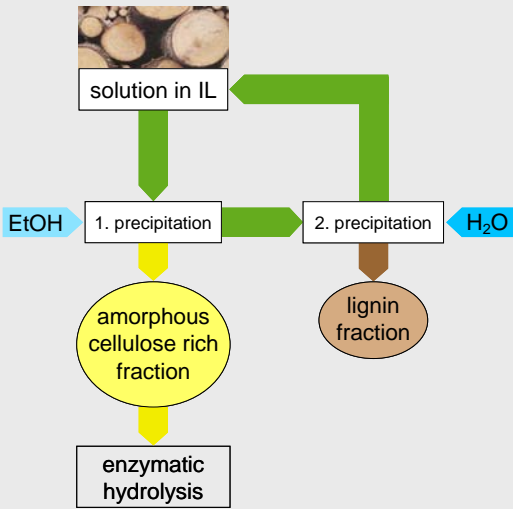
- liquid below 100 °C
- non flammable
- immiscible with many organic solvents
- BASF know-how & production
- various emerging applications
- dissolution of (ligno-)cellulose
- exclusive license from the University of Alabama (patents of Prof. Rogers)

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Example: biomass pretreatment

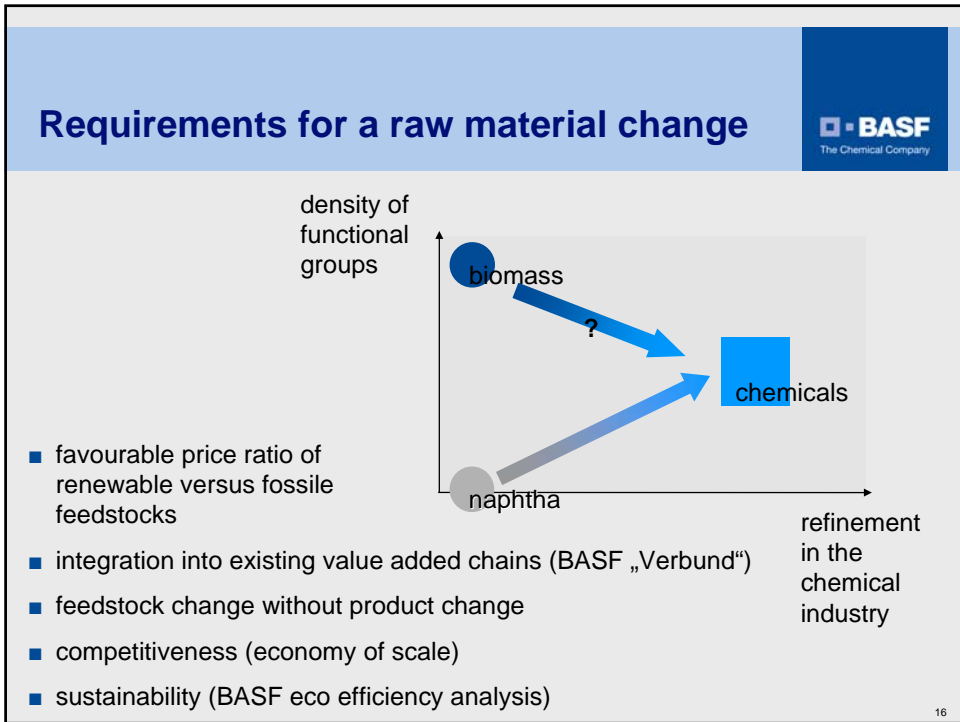
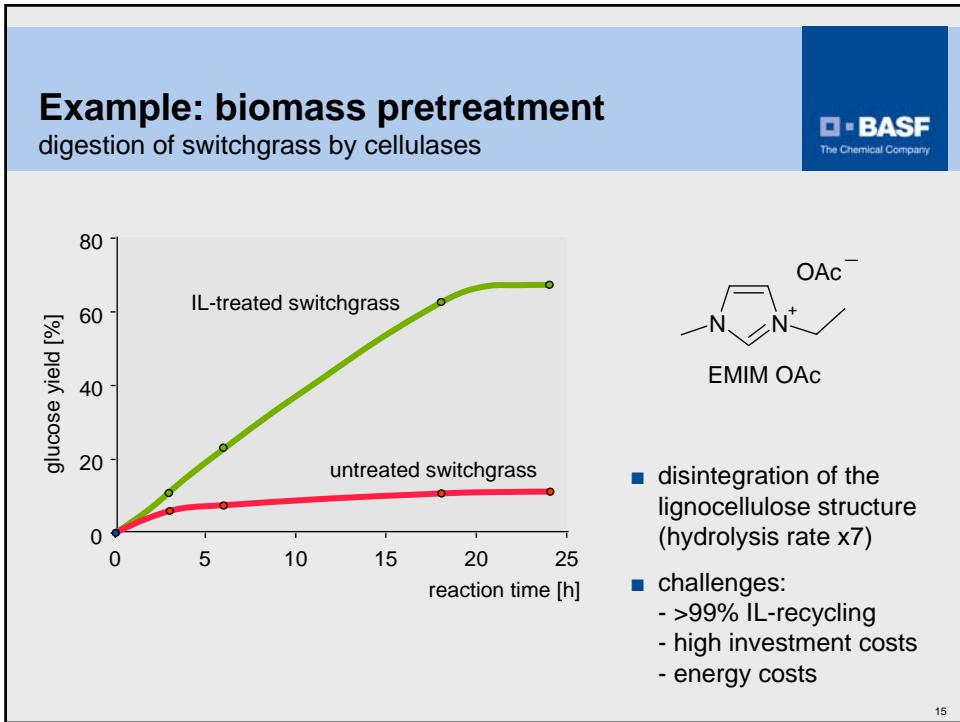
biorefinery with ionic liquids






- screening of >50 ILs
- screening of parameters:
 - temperature
 - precipitating agent
 - water content of the IL
 - precipitation protocol
- series of experiments in closed process cycles
- two patents filed:
 - WO 2008090155
 - WO 2008090156

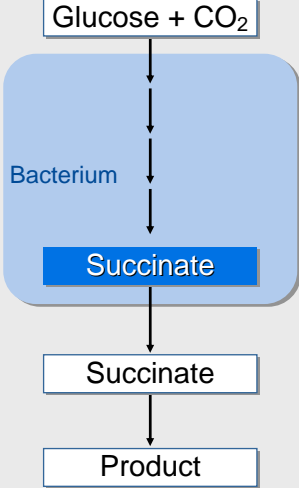
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Chemicals via Fermentation

Succinate as Intermediate and Monomer





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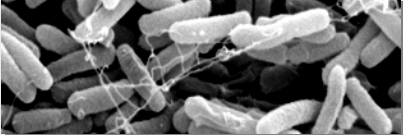
            graph TD
            A[Glucose + CO2] --> B[Bacterium]
            B --> C[Succinate]
            C --> D[Succinate]
            D --> E[Product]
            
```

Chances:

- Succinate as monomer and intermediate
- Potential for 100% yield
- CO₂ fixation

Challenges:

- Identify suitable production organism
- Improve microbial strain & fermentation
- Develop “*In broth chemistry*”




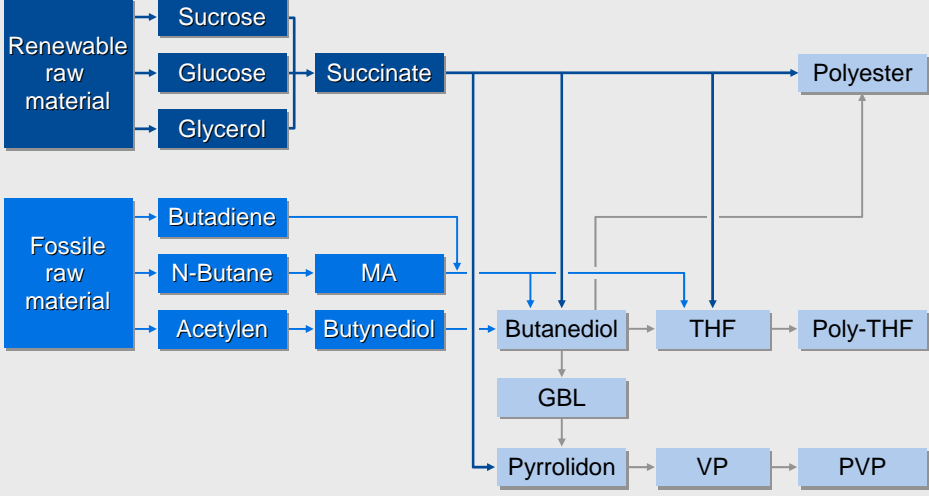
Succinate producing bacterium

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Chemicals via Fermentation

Succinate as Intermediate and Monomer






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            graph LR
            subgraph Renewable [Renewable raw material]
            R1[Sucrose]
            R2[Glucose]
            R3[Glycerol]
            end
            subgraph Fossil [Fossile raw material]
            F1[Butadiene]
            F2[N-Butane]
            F3[Acetylen]
            end
            R1 --> S[Succinate]
            R2 --> S
            R3 --> S
            F1 --> MA[MA]
            F2 --> MA
            F3 --> B[Butynediol]
            S --> P[Polyester]
            S --> B
            S --> THF
            S --> GBL
            S --> VP
            MA --> B
            B --> THF
            B --> GBL
            B --> VP
            THF --> PTHF[Poly-THF]
            GBL --> VP
            VP --> PVP
            
```

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Chemicals via Fermentation

Diaminopentane – New Monomer by Synthetic Metabolic Pathways



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Glucose

↓


Bacterium

↓

Lysine

↓

Lysine



Glucose

↓

Bacterium

↓

Lysine

⇓


1.5-Diaminopentane

↓


1.5-Diaminopentane

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Example: glycerol conversion



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glycerol

- epichlorohydrine
- propylene glycol
- acrylic acid
- glycerol carbonate
- glycerol oligomers

technical process
 pilot plant
 lab scale

Example: propylene glycol

alternative process based on glycerol



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The Chemical Company

- hydrogenation of glycerol

$$\text{HO-CH}_2\text{-CH(OH)-CH}_2\text{-OH} + \text{H}_2 \xrightarrow{\text{catalyst}} \text{HO-CH}_2\text{-CH(OH)-CH}_3 + \text{H}_2\text{O}$$


- challenges
 - volatile & rising glycerol price
 - reliable sourcing of >100.000 t/a difficult

(■ state of the art: epoxidation & hydrolysis of propylene)

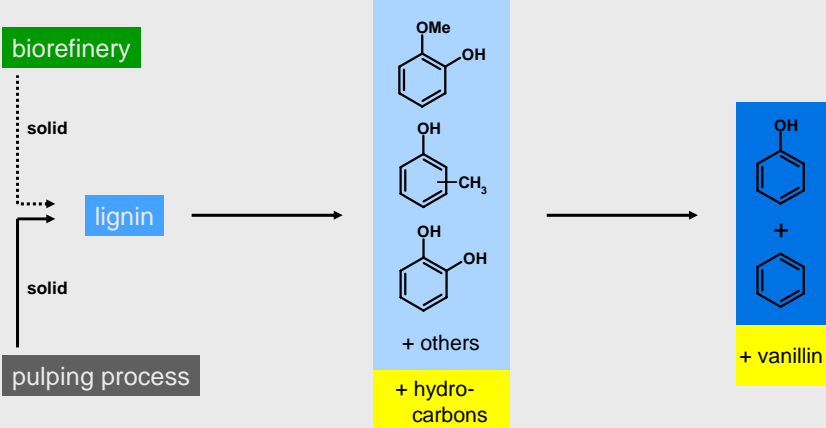
$$\text{CH}_2=\text{CH-CH}_3 \xrightarrow{\text{epoxidation}} \text{propylene oxide} \xrightarrow{\text{H}_2\text{O}} \text{HO-CH}_2\text{-CH(OH)-CH}_3$$

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Example: aromatics from lignin



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- biomass: 20-30% lignin
- max. yield of phenol/benzene = 45% from lignin

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Summary

Chemicals from renewable raw materials



- Renewable raw materials are well-established in the chemical industry
- Sufficient renewable raw materials must be available at competitive prices:
 - A rising price ratio of 'fossil to renewable raw materials' elevates the importance of renewable raw materials
 - An increasing raw material competition between chemical products based on renewable raw materials, biofuels and nutrition may be expected
- Cost effectiveness and technical feasibility of base chemicals from renewable raw materials have to be explored
- Sustainability has to be analyzed carefully for every alternative process or product (life cycle analysis, e.g. via BASF eco-efficiency analysis)
- Chemical products from renewable raw materials requires:
 - Verbund structure and value added chains based on renewable raw materials
 - A broad technology portfolio:
 - chemical catalysis: higher selectivity and stability
 - biotechnological catalysis: higher stability and space time yield
 - chemical engineering: solid handling and downstream

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